

Product Data Sheet

AMBERLITE™ IRC83 Ion Exchange Resin

High Capacity Weak Acid Cation Resin for Industrial Water Dealkalization and Oil Field Softening

Description

AMBERLITE™ IRC83 Ion Exchange Resin is a macroporous weak acid cation exchange resin designed for water utility engineers and oil field operators concerned with water and chemical efficiency. The distinct polymer structure of AMBERLITE IRC83 offers up to 10% more operating capacity versus other weak acid cation exchange resins reducing regeneration water use and waste volume. When used in conjunction with strong acid cation exchange resin, significant chemical savings can also be realized.

The exceptional physical and chemical stability coupled with the high operating capacity makes AMBERLITE IRC83 an excellent choice to tackle the demanding conditions for oil field softening (SAGD) systems. In addition to these performance benefits, users are realizing the long-lasting quality provided by AMBERLITE™ Ion Exchange Resins of over 60 years.

Typical Physical and Chemical Properties**

Matrix	Polyacrylic, macroporous		
Туре	Weak acid cation		
Functional Group	Carboxylic acid		
Physical Form	Off-white, opaque, spherical beads		
Ionic Form as Shipped	H ⁺ Form		
Total Exchange Capacity	≥ 4.60 eq/L		
Water Retention Capacity	40 – 50%		
Particle Size			
Particle Diameter ^b	500 – 750 μm		
Uniformity Coefficient	≤ 1.60		
< 300 µm	≤ 1.0%		
> 1180 µm	≤ 1.0%		
Swelling	$H^+ \rightarrow Na^+ : \le 60\%$		
Bulk Density, as Shipped ^c	700 g/L		

^b For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775). Cas per the backwashed and settled density of the resin, determined by ASTM D-2187.

Suggested Operating Conditions**

100°C (212°F)		
700 mm (2.3 ft)		
5 – 50 BV*/h		
See Figures 1 and 3		
2 – 4 BV/h		
6 – 40 BV/h		
2 – 4 BV/h		
2 BV at 2 – 4 BV/h		
2 – 4 BV at service flowrate		
HCI	H ₂ SO ₄	NaOH
2 – 5%	0.5 - 0.7%	2 – 4%
105 – 110%	105 – 110%	
110 – 160%	110 – 160%	110 – 160%
	700 mm (2.3 ft) 5 – 50 BV*/h See Figures 1 ar 2 – 4 BV/h 6 – 40 BV/h 2 – 4 BV/h 2 BV at 2 – 4 BV 2 – 4 BV at servi HCl 2 – 5% 105 – 110%	700 mm (2.3 ft) 5 – 50 BV*/h See Figures 1 and 3 2 – 4 BV/h 6 – 40 BV/h 2 – 4 BV/h 2 BV at 2 – 4 BV/h 2 – 4 BV at service flowrate HCl H ₂ SO ₄ 2 – 5% 0.5 – 0.7% 105 – 110% 105 – 110%

^{* 1} BV (Bed Volume) = 1 m³ solution per m³ resin or 7.5 gal per ft³ resin

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Hydraulic Characteristics

Bed expansion of AMBERLITETM IRC83 Ion Exchange Resin as a function of backwash flowrate and temperature ranging from $10-60^{\circ}\text{C}$ ($50-140^{\circ}\text{F}$) is shown in Figure 1 for resin in the H⁺ form and in Figure 3 for the Na⁺ form.

Pressure drop data for AMBERLITE IRC83 as a function of service flowrate and temperature ranging from 10 - 60°C (50 - 140°F) is shown in Figure 2 for resin in the H⁺ form and in Figure 4 for the Na⁺ form. Pressure drop data are valid at the start of the service run with clean water and a correctly classified bed.

Figure 1: Backwash Expansion (H⁺ form)

Temperature = $10 - 60^{\circ}\text{C} (50 - 140^{\circ}\text{F})$

18 20 10 12 14 16 140 120 100 % Bed Expansion 80 60 40 20 0 0 5 10 15 20 25 30 35 40 45 50 m/h Flowrate

Figure 2: Pressure Drop (H⁺ form)

Temperature = 10 - 60°C (50 - 140°F)

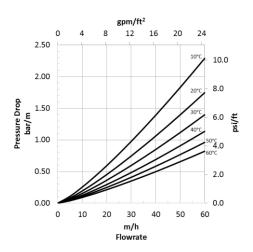


Figure 3: Backwash Expansion (Na⁺ form)

Temperature = $10 - 60^{\circ}\text{C} (50 - 140^{\circ}\text{F})$

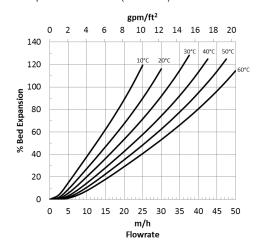
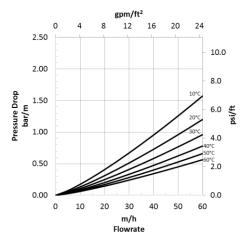


Figure 4: Pressure Drop (Na⁺ form)

Temperature = 10 - 60°C (50 - 140°F)



Application Information

Operating capacity

The operating capacity of AMBERLITE™ IRC83 Ion Exchange Resin is a function of water composition, temperature, and service flowrate. Contact your Dow representative if more specific information is required for your application.

Regeneration

AMBERLITE $^{\text{TM}}$ IRC83 Ion Exchange Resin is readily regenerated with as little as 1.05 times stoichiometric amounts of strong acids. If sulfuric acid is used, care must be taken to apply a low concentration of H_2SO_4 (0.7%) in order to minimize the potential of calcium sulfate precipitation.

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WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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